Vol. 2, No. 11

STABILITY OF CrO2

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Vol. 2, No. 11

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Pressure (kb)	Temp (°C)	Time (min)	Starting Material	Results
50	1800	2	CrO3	CrO <sub>2</sub>
25	1800	1	Cro3	CrO <sub>2</sub>
25	1900	1	CrO <sub>3</sub>	CrO <sub>2</sub>
50	1300	3	cro3	Cr0 <sub>2</sub>
50	1175	5	Cr <sub>2</sub> 0 <sub>5</sub>	CrO <sub>2</sub>
50	2000	2	CrO <sub>2</sub>	$cro_2$

The phases in the quenched material were identified by x-ray and optical (both reflected and transmitted light) observation and by a qualitative magnetic check for  ${\rm CrO}_2$ . The stoichiometry of the  ${\rm CrO}_2$  phase was checked by weight loss according to the reaction

$$2 \text{ CrO}_2 \rightarrow \text{Cr}_2\text{O}_3 + \frac{1}{2} \text{ O}_2$$

and the deviation from stoichiometry was found to be negligible.

## Results and Discussion

The results indicate that  ${\rm CrO}_2$  can be maintained without decomposition at temperatures up to  $2000^{\rm O}{\rm C}$  for short times (at least 2 minutes) and at temperatures to at least  $1200^{\rm O}{\rm C}$  for 60 minutes. Besides the characterization of  ${\rm CrO}_2$  structurally, chemically, and magnetically, another convincing proof for the stability of  ${\rm CrO}_2$  in the  $1200^{\rm O}\text{-}1550^{\rm O}{\rm C}$  temperature range was an increase in grain size of about one order of magnitude over that of the original approximately 5 $\mu$  to  $10\mu$  powder without the formation of  ${\rm Cr}_2{\rm O}_3$ .

Cr<sub>2</sub>O<sub>3</sub> Cr<sub>2</sub>O<sub>3</sub>

203